

**SYNTHESIS OF  
14(R,S)-[<sup>18</sup>F]FLUORO-6-THIA-HEPTADECANOIC ACID (FTHA)**

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**SUMMARY**

No-carrier-added (n.c.a) 14(R,S)-[<sup>18</sup>F]fluoro-6-thia-hepta-decanoic acid (FTHA) has been synthesized for evaluation as a PET tracer for myocardial long chain fatty acid utilization. The n.c.a. nucleophilic radiofluorination of benzyl 14(R,S)-tosyloxy-6-thia-heptadecanoate in acetonitrile utilized (cryptate 2.2.2/K)<sub>2</sub>CO<sub>3</sub> for anion activation. The resulting [<sup>18</sup>F]fluoro-ester was quantitatively hydrolyzed with addition of aqueous KOH and the product purified by reversed phase HPLC. The radiochemical yield of purified FTHA was 35-65% (n=5) with a synthesis time of 50 min. Radiochemical purity was >99%.

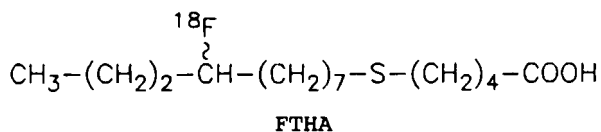
Key words: <sup>18</sup>F, nucleophilic radiofluorination, thia fatty acid

**INTRODUCTION**

Positron-emitter labeled long chain fatty acids (LCFAs) have been developed for non-invasive PET studies of LCFA utilization in the heart (for a review, see 1). [1-<sup>11</sup>C]Palmitate (CPA) has been extensively studied as a natural tracer of palmitate utilization. However, the interpretation of myocardial time-activity curves of CPA is complicated by the rapidity and multiplicity of clearance processes involved. The β-methyl substituted analog of palmitate, β-methyl-[1-<sup>11</sup>C]heptadecanoic acid has been proposed to provide a longer myocardial retention of the radiolabel as a consequence of inhibited β-oxidation (2). In light of the favorable properties of

$^{18}\text{F}$  ( $T_{1/2}=110$  min, low  $\beta^+$  energy) for PET imaging,  $^{18}\text{F}$ -labeled fatty acids were developed. The first were straight chain LCFA analogs with carrier [ $^{19}\text{F}$ ]fluoride added (3,4). No-carrier-added (n.c.a.) syntheses of 6- and 7- $^{18}\text{F}$ -labeled palmitate (5),  $\omega$ - $^{18}\text{F}$ -labeled straight chain (6-7), and  $\omega$ - $^{18}\text{F}$ -labeled methyl-substituted LCFA analogs have also been reported (8-10).

Recent biochemical studies show the CoA thioesters of 4-thia fatty acid analogs to be potent inhibitors of  $\beta$ -oxidation (11). Although the mechanism of inhibition is not known,  $\beta$ -oxidation of 4-thia fatty acids may proceed to 4-thia-2-enoyl-CoA thioesters which could possibly bind to acyl-CoA dehydrogenases within the mitochondrion (cf. 12). To determine whether an  $^{18}\text{F}$ -labeled, even-substituted thia LCFA analog could be metabolically trapped in myocardium, we have synthesized 14(*R,S*)- $^{18}\text{F}$ fluoro-6-thia-heptadecanoic acid (FTHA) in no-carrier-added form.



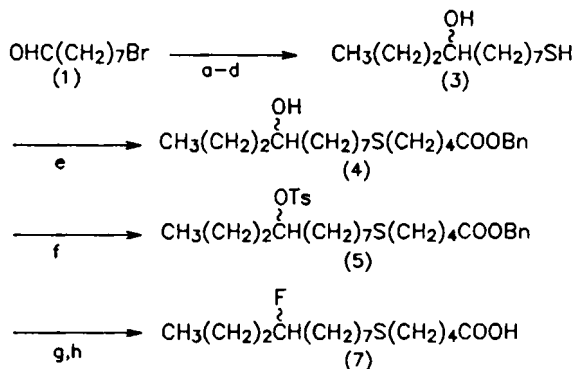
#### RESULTS AND DISCUSSION

The n.c.a. nucleophilic radiofluorination of benzyl 14(*R,S*)-tosyloxy-6-thia-heptadecanoate **5** in acetonitrile utilized (cryptate 2.2.2/K) $_2\text{CO}_3$  for anion activation (7). Incorporation of [ $^{18}\text{F}$ ]fluoride was 40-82%. The resulting [ $^{18}\text{F}$ ]fluoro-ester (FTHA-Bn) was quantitatively hydrolyzed with the addition of aqueous KOH and purified by reversed-phase HPLC (Table 1). Radiochemical yield of purified FTHA was 35-65% ( $n=5$ ) with a synthesis time of 50 min.

Other than [ $^{18}\text{F}$ ]fluoride, only minor radiochemical impurities (< 5%) were observed in the HPLC radiochromatograms. A large fraction of the unincorporated [ $^{18}\text{F}$ ]fluoride was retained on the HPLC column, the remainder eluting with the void volume or bleeding slowly from the column. For experiments that were sensitive to the small [ $^{18}\text{F}$ ]fluoride impurity, the product fraction was passed through a silica gel SEP-PAK cartridge (400 mg, Millipore) before evaporation of solvent. This reduced the contamination of [ $^{18}\text{F}$ ]fluoride typically from 0.2% to <.03%. The radiochemical purity of FTHA, as monitored up to 4 hr after synthesis by

analytical reversed-phase HPLC, was >99%.

The tosyloxy-ester **5** was prepared in several synthetic steps according to Scheme 1. This compound was an oil and stable when refrigerated. Non-radioactive standards of fluorinated fatty acid **7** and benzyl ester **6** were prepared via fluorination of **5** with tetraethylammonium fluoride.



a)  $\text{CH}_3(\text{CH}_2)_2\text{MgBr}$ , THF; b)  $\text{H}^+$ ; c)  $(\text{NH}_2)_2\text{CS}$ , DMSO; d) aq. KOH,  $\Delta$ ;  
 e)  $\text{Br}(\text{CH}_2)_4\text{COOBn}$ ,  $\text{K}_2\text{CO}_3$ , DMF; f)  $(\text{Ts})_2\text{O}$ ,  $\text{C}_5\text{H}_5\text{N}$ ;  
 g) TEAF,  $\text{CH}_3\text{CN}$ ; h) KOH, EtOH/ $\text{H}_2\text{O}$

**Scheme 1**

## EXPERIMENTAL

### General

Chemicals were of analytical grade unless otherwise noted. DMF and DMSO were stored over molecular sieves. THF was dried by distillation from  $\text{LiAlH}_4$ . Dry  $\text{CH}_3\text{CN}$  (<0.003%  $\text{H}_2\text{O}$ ) was obtained commercially (Merck).  $^1\text{H-NMR}$  spectra were recorded with a Varian Model-390 90 MHz spectrometer using  $\text{CDCl}_3$  as solvent ( $\text{Me}_4\text{Si}$ , 0.00 ppm).  $R_f$  values refer to thin layer chromatography (TLC) performed on silica gel (Merck) with the solvent system noted. Routine column chromatography was performed under normal pressure with silica gel (Merck, 70-325 mesh).

### 11-Bromo-4(R,S)-undecanol (2)

8-Bromo-1-octanal **1** was prepared in 76% yield from 8-bromo-1-octanol according to the method of Corey et al. (13). The aldehyde was isolated by column chromatography (hexane/ether 3:1). TLC (hexane/ether 3:1)  $R_f=0.7$ . IR  $1730\text{ cm}^{-1}$ . The aldehyde was allowed to react with propyl magnesium bromide in dry THF in the classical

manner for a Grignard reaction. Hydrolysis of the resulting magnesium alkoxide was performed by the slow addition of ice-diluted HCl. The organic phase was separated, dried over  $\text{Na}_2\text{SO}_4$ , and evaporated under reduced pressure. The product 2 was isolated as an oil by column chromatography (hexane/ether 1:2) in 95% yield. TLC (hexane/ether 1:2)  $R_f=0.6$ . NMR 0.94 (t, 3H, C(1) $\text{H}_3$ ), 1.40 (m, 16H,  $\text{CH}_2$ ), 3.5 (m, 3H, C(11) $\text{H}_2$ , C(4)H). Anal.- Calc. for  $\text{C}_{11}\text{H}_{23}\text{BrO}$ : C, 52.59; H, 9.23; Br, 31.8. Found: C, 53.23; H, 9.22; Br, 32.0.

#### 11-Mercapto-4(R,S)-undecanol (3)

To a solution of 2 (13 g, 52 mmol) in DMSO (15 ml) at 24 °C was added thiourea (55 mmol). After 20 hr, 2N KOH (10 ml) was added and the mixture was heated at 80 °C for 5 min. The mixture was acidified (HCl) and extracted twice with ether (40 ml). The combined ether phases were washed successively with water and brine, dried over  $\text{Na}_2\text{SO}_4$ , and evaporated under reduced pressure. The product 3 (7 g, 66% yield) was isolated as a colorless oil by column chromatography (hexane/ether 1:3). TLC (hexane/ether 1:3)  $R_f=0.65$ . NMR 0.94 (t, 3H, C(1) $\text{H}_3$ ), 1.40 (m, 16H,  $\text{CH}_2$ ), 2.51 (quartet, 2H, C(11) $\text{H}_2$ ), 3.64 (m, 1H, C(4)H). Anal.- Calc. for  $\text{C}_{11}\text{H}_{24}\text{OS}$ : C, 64.65; H, 11.84; O, 7.83; S, 15.7. Found: C, 64.10; H, 12.08; O, 7.82; S, 15.6.

#### Benzyl 14(R,S)-hydroxy-6-thia-heptadecanoate (4)

To a solution of benzyl 5-bromopentanoate (2.7 g, 10 mmol, ref. 14) in dry DMF (100 ml) at 24 °C was added the mercaptan 3 (2.3 g, 9.9 mmol) and  $\text{K}_2\text{CO}_3$  (0.7 g, 5 mmol). After reaction under argon for 20 hr, the mixture was acidified with ice-diluted HCl and extracted with ether (2 x 50 ml). The combined ether fractions were washed successively with dilute  $\text{NaHCO}_3$ , water and brine, dried over  $\text{Na}_2\text{SO}_4$ , and evaporated under reduced pressure. The product 4 (1.6 g, 40% yield) was isolated by column chromatography (hexane/ether 1:2) and recrystallized from hexane to give white flakes. M.p.= 31.7 °C. TLC (hexane/ether 1:2)  $R_f=0.5$ . NMR 0.92 (t, 3H, C(17) $\text{H}_3$ ), 1.40 (m, 20H,  $\text{CH}_2$ ), 2.4 (m, 6H, C(2) $\text{H}_2$ , C(5) $\text{H}_2$ , C(7) $\text{H}_2$ ), 3.69 (m, 1H, C(14)H), 5.10 (s, 2H,  $-\text{CH}_2-\phi$ ), 7.32 (s, 5H, aryl). Anal.- Calc. for  $\text{C}_{23}\text{H}_{38}\text{O}_3\text{S}$ : C, 70.01; H, 9.71; O, 12.2; S, 8.12. Found: C, 69.65; H, 9.76; O, 12.1; S, 8.07.

#### Benzyl 14(R,S)-tosyloxy-6-thia-heptadecanoate (5)

The hydroxy-ester 4 (0.44 g, 1.11 mmol) was dissolved in  $\text{CH}_2\text{Cl}_2$  (5 ml) at 0 °C. Pyridine (1.4 mmol) and p-toluenesulfonic

anhydride (1.4 mmol) were added and the mixture was allowed to react for 4 hr. The mixture was diluted with ether (10 ml) and washed successively with dil. HCl, dil. NaHCO<sub>3</sub>, water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and the solvent evaporated under reduced pressure. The product **5** (0.56 g, 92% yield) was isolated by column chromatography (hexane/ethyl acetate 3:1), giving a colorless oil. TLC (hexane/ethyl acetate 3:1) R<sub>f</sub>=0.6. NMR 0.92 (t, 3H, C(17)H<sub>3</sub>), 1.40 (m, 20H, CH<sub>2</sub>), 2.4 (m, 9H, C(2)H<sub>2</sub>, C(5)H<sub>2</sub>, C(7)H<sub>2</sub>, (tosyl) CH<sub>3</sub>), 4.57 (m, 1H, C(14)H), 5.09 (s, 2H, -CH<sub>2</sub>-φ), 7.5 (m, 9H, aryl). Anal.- Calc. for C<sub>30</sub>H<sub>44</sub>O<sub>5</sub>S<sub>2</sub>: C, 65.66; H, 8.08. Found: C, 65.28; H, 8.08.

#### Benzyl 14(R,S)-fluoro-6-thia-heptadecanoate (6)

Tetraethylammonium fluoride hydrate (5 mmol) was dried by azeotropic distillation with dry CH<sub>3</sub>CN (2 x 20 ml). A solution of the tosylate **5** (1 mmol) in dry CH<sub>3</sub>CN (10 ml) was added and the mixture was allowed to react at 24 °C for 20 hr. After evaporation of the solvent, water (20 ml) and ether (20 ml) were added. The organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent evaporated. The product **6** (24% yield) was isolated by HPLC (Table 1). TLC (hexane/ethyl acetate 3:1) R<sub>f</sub>=0.8. NMR 0.92 (t, 3H, C(17)H<sub>3</sub>), 1.40 (m, 20H, CH<sub>2</sub>), 2.45 (m, 6H, C(2)H<sub>2</sub>, C(5)H<sub>2</sub>, C(7)H<sub>2</sub>), 4.11 (m, 0.5H, C(14)H), 4.75 (m, 0.5H, C(14)H), 5.09 (s, 2H, -CH<sub>2</sub>-φ), 7.32 (s, 5H, aryl).

#### 14(R,S)-Fluoro-6-thia-heptadecanoic acid (7)

The fluoro-ester **6** (1 mmol) was hydrolyzed by treatment with 15 ml EtOH/0.2N aq. KOH (1:1) at 24 °C for 20 hr. After acidification (HCl), the crude product was extracted into ether (25 ml). The ether phase was dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent evaporated under reduced pressure. The product **7** was isolated by preparative HPLC (Table 1) and recrystallized from hexane. M.p.= 50.3 °C. TLC (hexane/ethyl acetate/acetic acid 75:25:2) R<sub>f</sub>=0.35. NMR 0.92 (t, 3H, C(17)H<sub>3</sub>), 1.40 (m, 20H, CH<sub>2</sub>), 2.45 (m, 6H, C(2)H<sub>2</sub>, C(5)H<sub>2</sub>, C(7)H<sub>2</sub>), 4.13 (m, 0.5H, C(14)H), 4.75 (m, 0.5H, C(14)H).

#### <sup>18</sup>F-labeling procedure

To a 20 ml glassy carbon vessel was added Kryptofix 2.2.2 (10 mg, 25 μmol), CH<sub>3</sub>CN (0.5 ml) and 20 μl of a 9% K<sub>2</sub>CO<sub>3</sub> solution in water. [<sup>18</sup>F]Fluoride, produced via proton bombardment of H<sub>2</sub><sup>18</sup>O (>95 atom %), was then added, the vessel placed into a 90-95 °C oil bath, and the solvent evaporated under a stream of helium. The residue was

**Table 1.** Preparative reverse-phase HPLC capacity factors ( $k'$ ) of LCFA analogs and esters (Nucleosil, C-18 (10 $\mu$ ), 250 x 20 mm; MeOH/H<sub>2</sub>O/ACOH (88:12:0.4); flow: 10 ml/min)

Compound	$k'$
Benzyl 14-tosyloxy-6-thia-heptadecanoate (5)	3.2
Benzyl 14-fluoro-6-thia-heptadecanoate (6, FTHA-Bn)	3.5
14-Tosyloxy-6-thia-heptadecanoic acid	2.1
14-Fluoro-6-thia-heptadecanoic acid (7, FTHA)	2.4

further dried by azeotropic distillation with CH<sub>3</sub>CN (2 x 0.3 ml). A solution of the benzyl ester 5 (25  $\mu$ mol) in dry acetonitrile (0.5 ml) was added and a water-cooled condenser was placed atop the reaction vessel. Reaction time was 15 min. The incorporation of [<sup>18</sup>F]fluoride was monitored by radio-TLC (hexane/ethyl acetate 3:1). R<sub>f</sub> values were 0.0 and 0.8 for [<sup>18</sup>F]fluoride and [<sup>18</sup>F]fluoro-ester, respectively.

Subsequent hydrolysis of the resulting [<sup>18</sup>F]fluoro-ester was performed in the same vessel by the addition of 0.3 ml 0.2N KOH and continued heating at 90-95 °C for 5 min. The mixture was cooled, acidified with concentrated acetic acid (30 $\mu$ l), and applied to the preparative HPLC column (Table 1). After evaporation of solvent under reduced pressure, the residue was typically formulated in a 0.9% NaCl solution containing 4-6% albumin for *in vivo* studies in small animals. Alternatively, FTHA may be formulated in patient's serum for clinical use (7).

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